SCORE Water Quality Monitoring Program

- Sponsored by SCDNR
 - Co-sponsor: Friends of the Rivers
- Open to any interested citizens
- Training and equipment provided
- Data collected is available to all on our website
- Data supplements that collected by SCDNR

Why Does SCORE Monitor Water Quality?

- Document water quality effects on restoration
 - Explain differences among reefs
 - Determine factors which make a "good" site
 - Develop site selection decision matrix
 - Refine site selection by pre-construction monitoring
- Document restoration effects on water quality

Monitor Responsibilities

- •Reliability
 - •If you cannot make a scheduled monitoring session, notify the coordinator or find a substitute
- •Careful observation and recording
- •Instrument calibration
- •Care/cleaning of instruments
- •Return kit to central location after use
- •Report problems to us or Friends of Rivers
- •Enter data (or send to us)

When to monitor

- Weekly
 - If we have 4 teams per site, each team goes once a month
- Daylight noon is best
- Tide any
- Year round
- Not during rain
- Increased frequency after rain events

Where to monitor

Sites in Beaufort County

- Haig Landing, Pinckney Island
- Trask Landing, Sawmill Creek Road, Bluffton
- Sugar Mill Dock, Callawassie Island
- The Sands Landing, Port Royal
- Dataw Island Marina

Where is the equipment?

Waddell Mariculture Center, Sawmill Creek Road Discovery Museum, Hilton Head LowCountry Institute, Spring Island LowCountry Estuarium, Port Royal Dataw Marina

What is monitored

- Weather conditions
- Water clarity
- Temperature (air and water)
- Dissolved oxygen
- Salinity
- pH

Plan your trip

• Go during your assigned week

Take:

- Distilled water (available at grocery store for \$1)
- Empty plastic drink bottle
- Paper towels or towel
- Writing instrument, pad of paper
- Pick up the water quality kit
- Check the kit for all supplies before leaving
- Calibrate the pH probe

Kit contents

The following items should be in each kit:

- Data notebook with blank datasheets(should be in a ziplock bag)
- Plastic beaker (for sample collection)
- 1 liter bottle of distilled water (refill each time)
- Small drink bottle to use for disposal of glass (replace as needed)
- Pencils (replace if needed)
- pH probe with small screwdriver and spare batteries
- 500 ml bottle of pH 7.0 buffer
- 500 ml bottle of pH 10.0 buffer
- Two 100 ml plastic cups with lids
- Salinity refractometer with plastic pipet
- Thermometer with string and float
- Dissolved oxygen kit (in a black box)
- Secchi disk with calibrated rope
- Towel or paper towels (replace as needed)

pH Probe Calibration Pointers

- Only fill the cup about halfway
- If the pen does not stabilize or will not calibrate, try soaking in tap water for 5 minutes and repeat the calibration
- If the pen still will not calibrate, follow the troubleshooting instructions
- If the pen still will not calibrate, do not read pH that day, leave a note in the kit that the pen is not working, and notify us as soon as possible so that we can replace the pen

Weather Conditions

- Day and time allows us to determine exact tidal height
- Wind affects dissolved oxygen and water clarity
- Cloud cover affects water clarity
- Rainfall affects many parameters
- Tidal height affects salinity, temperature, dissolved oxygen, water clarity
- Current speed may affect water clarity and dissolved oxygen
- Water conditions may alert us to a special situation

Temperature

- Affects the amount of oxygen which can be dissolved in the water
- Affects metabolic rates
- Affects algal growth rates
- Promotes chemical reactions
- Affects sensitivity of organisms to disease, parasites and pollutants

Temperature

In addition to normal seasonal variations these factors may affect temperature:

- Industrial discharges
- Runoff from warm paving
- Suspended solids (from runoff)
- Removal of shoreline vegetation (reduces shade)

Temperature Pointers

- Air temperature must be measured with a dry thermometer (do this first)
- Make sure the thermometer is not in direct sunlight
- Allow ~ 5 minutes to stabilize
- Water temperature can be measured in the creek itself or in a sample taken with the 1 liter beaker
- If using a sample, measure temperature immediately, before the sample has a chance to warm up (or chill down)

Water Clarity

What is water clarity

- Ability of light to penetrate water
- Indirect measure of suspended solids in the water
- Water clarity is also affected by algae in the water Importance of water clarity
- Suspended solids may result in low dissolved oxygen
- Suspended solids settle out on reefs, smothering oysters
- Suspended solids block sunlight, limiting phytoplankton
- Suspended solids can interfere with oyster feeding
- Suspended solids may indicate runoff which may have other negative effects

Water Clarity

Factors which affect water clarity

- Runoff and erosion
- Excess nutrients (e.g. from fertilizers) can cause algal blooms which decrease water clarity
- Dying algae decreases water clarity and dissolved oxygen
- Wind and currents can stir up bottom sediments and keep particles in suspension
- Water clarity is greater at lower temperatures because particles do not stay in suspension as well
- Oyster reefs may affect water clarity by physical or biological removal of particles
- Oyster reefs may reduce erosion and related suspended solids

Secchi Disk Pointers

- Do not wear sunglasses!
- Measure on the shaded side of the dock (reduces glare off water) with your back to the sun
- Tie the line to your wrist so you won't lose the disk
- If the disk hits the bottom, record depth as > whatever mark you are at
- A clothespin or other clip may be used to mark the line to facilitate reading
- If possible, rinse the secchi disk and rope with fresh water after use
- Dry the secchi disk as much as possible before storing. It may be useful to keep it in a plastic bag

Water Sample Collection Pointers

- Rinse the 1 liter beaker with the sample water
- Tip the beaker sideways and submerge slowly to approximately 1 foot depth (up to your elbow)
- Turn the beaker upright and slowly remove from the water
- Take care not to splash which will introduce oxygen

Dissolved oxygen

- Amount of oxygen dissolved in the water
- Most critical parameter for aquatic life
- Can come from air (wind, waves) and from algae
- Affected by many other parameters
 - temperature
 - suspended solids
 - plant growth
 - excess nutrients (fertilizer)
 - chemical reactions
 - removal of shoreline vegetation (increases erosion and raises temperature)
- Oysters may indirectly effect dissolved oxygen by removal of algae and suspended solids

Dissolved oxygen pointers

- The sample ampoule and the standards should be dry when reading
- Remove the comparator from the box to use
- Always start your comparisons at the low end of the scale and work up
- Dispose of ampoule in empty soda bottle now safe to put in any trash receptacle
- Replace the soda bottle as needed
- Normal range of DO for our sites is 4 to 10. Readings will be highest in the winter.

Salinity

- Measure of the salts in water
- Salinity is needed to interpret dissolved oxygen
- Most animals have preferred salinity ranges. Oysters occur in 10 ppt to 35 ppt but prefer 25 ppt
- Some oyster diseases are related to salinity
- Salinity affects algal growth (food for oysters)
- Freshwater runoff will lower salinity
- Runoff may be accompanied by pollutants, solids, etc

Salinity Pointers

- Measured with a refractometer in parts per thousand (ppt)
- Refractometer must be calibrated with distilled water
- Do not submerse the refractometer
- Do not drop the refractometer
- If you do not get a sharp differential, try loading the refractometer again
- Rinse only the lens after use and pat dry (do not rub)
- Salinity will likely be in the range of 20 to 35ppt at our sites but may be lower after rainfall

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- Measure of the acidity or alkalinity of water
- 7.0 is neutral: greater than 7 is basic and lower than 7 is acid
- Logarithmic scale: a pH of 6.0 is ten times more acidic than 7.0
- Most organisms have narrow range of optimal pH
- pH effects rate of reactions
- Chemical pollutants affect pH
- Algal blooms affect pH these can be cause by excess nutrients in runoff or by sewage outfalls

pH Pen Pointers

- Normal seawater has a pH of ~8. pH will be lower if salinity is lower. Typical ranges you will see are 7.0 to 8.5
- The pen must be calibrated before use
- If the pen will not calibrate follow the troubleshooting procedure in your manual
- Do not record pH if the pen is not calibrating properly. Note on the sheet that it would not calibrate.
- Notify us as soon as possible if the pen is not working
- The upper part of the pen is very sensitive to water do not submerge above the connector seam.
- The cap must be removed before use and replaced after use.
- Place a few drops of the pH 7 buffer in the cap before you replace it to keep the probe moist.

Data Entry

- Website :http://www3.csc.noaa.gov/scoysters
- Path to data entry: click on Monitoring Data in upper left
- Entry options: Choose enter data online for water quality
- Choose your site
- Enter the data
- Click on review and make sure your data is correct. If not, click edit and make changes.
- When you are satisfied the data is correct, click on Submit
- If you notice a mistake after you submit the data, notify us. Do not reenter as this will just add another record.



For additional volunteer information, contact Nancy Hadley at <u>hadleyn@mrd.dnr.state.sc.us</u> or (843) 953-9841, or Michael Hodges at <u>hodgesm@mrd.dnr.state.sc.us</u> or (843) 953-9841.

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Quick Links







Go to "http://www3.csc.noaa.gov/scoysters/html/glossary.htm"

Project Brochure



spat, and survival and growth of juvenile oysters. This information helps scientists to understand how and why oyster reefs form and survive, and to evaluate the restoration techniques used. So, how are the sites doing so far? Follow the links below to view the data for a particular reef site or to compare data from multiple sites.





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Examining the Data

Click on Monitoring Data

Scroll to bottom of page and choose view data

You have many options (one site, multiple sites; one parameter, multiple parameters; graph or table)

This area of the website is being updated so it may change



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	Show all data for all water quality parameters.			
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	Choose the year(s) for which you would like to see	water quality data.		
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Oyste	er Recruitment and Growth			
	Compare Oyster Recruitment and Growth data from S	pring 2002 and 2003 Assessi	nents	
Note	Selecting all items can result in a large web page. Please b	e mindful of this if you are on	a slower connection.	
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Quality Assurance and Quality Control

- All volunteers are trained
- Instruments (pH probe, refractometer) should be calibrated each week
- DNR periodically checks the instruments against our electronic instruments
- DNR staff examine data monthly and note anomalies
- Volunteers may be contacted to determine cause of anomalies
- Retraining will be conducted if necessary
- Suspect data is deleted
- Notify us any time you have a doubt about your instruments

What we do with the data

- Compare water quality against oyster performance
- Watch for changes in water quality as oyster reefs mature
- Watch for adverse conditions that may harm oysters
- Learn what factors to look for in selecting a site